



(21)(A1) **2,253,695**
(22) 1998/11/30
(43) 1999/06/01

(72) KOTHRADE, Stephan, DE
(72) BERNDL, Gunther, DE
(72) ERNST, Andreas, DE
(72) SANNER, Axel, DE
(71) BASF AKTIENGESELLSCHAFT, DE
(51) Int.Cl.⁶ A61K 47/34, A61K 7/46, A61K 9/20, A61K 9/16, A23K 1/16,
A01N 25/10, A23L 1/03
(30) 1997/12/01 (19753300.0) DE
(54) PRODUCTION DE FORMES DE DOSAGE SOLIDES
(54) THE PRODUCTION OF SOLID DOSAGE FORMS

(57) Solid dosage forms are produced by mixing at least one polymeric binder, at least one active ingredient and, where appropriate, conventional additives, to form a plastic mixture, by using a homo- or copolymer of N-vinylcaprolactam as polymeric binder. It is possible in this way to produce solid dosage forms with rapid release of the active ingredient in a simple and low-cost manner.



The production of solid dosage forms

The invention relates to a process for producing solid dosage forms by mixing at least one polymeric binder, at least one active ingredient and, where appropriate, conventional additives to form a plastic mixture, and shaping the mixture. The invention particularly relates to a process for producing solid pharmaceutical dosage forms.

10

Classical processes for producing solid pharmaceutical forms, especially tablets, are carried out batchwise and comprise a plurality of stages. Pharmaceutical granules represent an important intermediate therefor. Thus, for example, Bauer, Frömmig and Führer, "Pharmazeutische Technologie", Georg-Thieme-Verlag, pages 292 et seq., reveal that drug forms can be obtained from the melt by dry granulation. The possibility of producing solidified melt granules either by melting and shock solidification, by casting and comminuting or by prilling in spray towers is described. One problem with these processes is the accurate shaping which is necessary for producing drugs. Irregular particles or fragments are produced, so that the resulting shape by no means corresponds to customary drug forms, and granules therefore have only little importance as a drug form on their own. Production of desired solid drug forms requires the use of further process steps such as compression in tabletting machines. This is time-consuming and costly.

A considerably simpler continuous process for producing solid pharmaceutical forms has been known for some time and entails extruding a solvent-free melt of a polymeric binder containing active ingredients, and shaping the extrudate to the required drug form, for example in a calender with molding rolls, see EP-A-240 904, EP-A-240 906, EP-A-337 256 and EP-A-358 105 (melt extrusion). It is possible in this way to achieve specific shaping. The polymeric binders employed are, in particular, polymers of N-vinylpyrrolidone or copolymers thereof, e.g. with vinyl acetate.

Homo- and copolymers of N-vinylcaprolactam have been used in conventional pharmaceutical dosage forms. Thus, US 3,900,559 describes a dosage form for the sustained release of methantheline from a matrix which comprises a water-insoluble, gel-forming, hydrophilic terpolymer with N-vinyl lactam as one constituent. US 4,987,182 describes the production of a mixture of polyvinylbutyral and a polyvinyl lactam which is used as active ingredient carrier. WO 96/17579 describes a binder for tablets

which comprises a homo- or copolymer of an N-vinyllactam and a polyacid. EP 756 820 A describes iodine-containing preparations with improved stability which may comprise polyvinylcaprolactam in the binder component. DE 44 34 986 A describes the preparation 5 of polyvinylcaprolactam. This is employed for producing slow-release active ingredient preparations. DE 43 27514 A discloses a copolymer of a water-insoluble monomer and a water-soluble monomer which may be N-vinylcaprolactam. The copolymer can be used as carrier for active ingredients and makes easier 10 incorporation of the components and controlled release possible.

DE 44 22 881 describes the use of copolymers of N-vinylcaprolactam with hydrophobic comonomers in formulations of active ingredients with low water solubility. The formulations are in 15 the form of solutions or dispersions and have improved stability and efficacy.

DE 35 17 080 describes a component for therapeutic active ingredient delivery systems based on polymers which have 20 elastomeric properties and may be homo- or copolymers of N-vinyllactams. These components are used in the active ingredient reservoir layer for therapeutic active ingredient delivery systems, for example medicinal plasters.

25 Dosage forms based on polymers of this type have the disadvantage that they release the active ingredient too slowly for many applications. It is therefore impossible to produce rapid-release dosage forms, also called instant release formulations, without taking additional measures, such as comminution of the dosage 30 form, to produce a large surface area.

It is an object of the present invention to provide dosage forms which can be produced by melt extrusion and are capable of rapid release of active ingredient.

35 We have found that this object is achieved by using an N-vinyl-caprolactam-containing polymer as polymeric binder.

The present invention therefore relates to a process for 40 producing solid dosage forms by mixing at least one polymeric binder, at least one active ingredient and, where appropriate, conventional additives to form a plastic mixture, and shaping, wherein a homo- or copolymer of N-vinylcaprolactam is used as polymeric binder.

45 The novel process makes it possible to produce solid dosage forms with rapid release of active ingredient ("instant release") in a

simple and low-cost manner. The rapid release of the active ingredient is surprising because N-vinylcaprolactam polymers in the polymer art release the active ingredient only slowly, and because plastic mixtures of the components result in very 5 compact, nonporous dosage forms with a small surface area, which would suggest slow release of active ingredient.

Dosage forms mean herein all forms which are suitable for use as drugs, plant treatment compositions, human and animal foods and 10 for delivering fragrances and perfume oils. These include, for example, tablets of any shape, pellets, granules, but also larger forms such as cubes, blocks (bricks) or cylindrical forms, which can be used, in particular, as human or animal foods.

15 The dosage forms obtainable according to the invention generally comprise:

- a) 0.1-99% by weight, in particular 0.1-60% by weight (based on the total weight of the dosage form) of an active ingredient,
- 20 b) 10-99.9% by weight, in particular 40-99.9% by weight of a polymeric binder and
- c) where appropriate additives.

25 The polymeric binder employed can be a homopolymer or a copolymer of N-vinylcaprolactam. The copolymer contains at least 1% by weight, preferably at least 10% by weight, particularly preferably at least 25% by weight and, in particular, at least 30 50% by weight of N-vinylcaprolactam units.

As copolymer, the polymeric binder contains at least one other copolymerizable monomer in an amount of from 0.5 to 99% by weight, preferably 0.5 to 90% by weight, particularly preferably 35 0.5 to 75% by weight and, in particular, 0.5 to 50% by weight. Suitable comonomers are, in particular, monoethylenically unsaturated carboxylic acids having 3 to 30, in particular 3 to 40 8, carbon atoms, such as acrylic acid, methacrylic acid, dimethylacrylic acid, ethacrylic acid, maleic acid, citraconic acid, methylenemalonic acid, allylactic acid, vinylacetic acid, crotonic acid, fumaric acid, mesaconic acid and itaconic acid, and the monoesters of said dicarboxylic acids with C₁-C₂₄-alkanols. Acrylic acid, methacrylic acid, maleic acid or mixtures of said carboxylic acid are preferred. The monoethylenically 45 unsaturated carboxylic acids may be employed in the form of the free acid and, if available, the anhydrides or in partially or completely neutralized form. The neutralization is preferably

carried out using alkali metal or alkaline earth metal bases, ammonia or amines, eg. sodium hydroxide, potassium hydroxide, sodium carbonate, potassium carbonate, sodium bicarbonate, magnesium oxide, calcium hydroxide, calcium oxide, gaseous or aqueous ammonia, triethylamine, ethanolamine, diethanolamine, morpholine, diethylenetriamine or tetraethylenepentamine.

Further examples of suitable comonomers are the esters of the abovementioned carboxylic acids with C₁-C₂₄-alkanols, C₂-C₄-diols, mono- and di-C₁-C₄-alkylamino-C₂-C₄-alkanols, and the amides, mono- and di-C₁-C₄-alkylamides and nitriles of these carboxylic acids, eg. methyl acrylate, ethyl acrylate, methyl methacrylate, ethyl methacrylate, hydroxyethyl acrylate, hydroxypropyl acrylate, hydroxybutyl acrylate, hydroxyethyl methacrylate, hydroxypropyl methacrylate, hydroxyisobutyl acrylate, hydroxyisobutyl methacrylate, monomethyl maleate, dimethyl maleate, monoethyl maleate, diethyl maleate, 2-ethylhexyl acrylate, 2-ethylhexyl methacrylate, stearyl acrylate, stearyl methacrylate, behenyl acrylate, behenyl methacrylate, octyl acrylate, octyl methacrylate, acrylamide, methacrylamide, N,N-dimethylacrylamide, N-tert-butyl acrylamide, acrylonitrile, methacrylonitrile, dimethylaminoethyl acrylate, diethylaminoethyl acrylate, diethylaminoethyl methacrylate and the salts of the last-mentioned monomers with carboxylic acids or mineral acids or the quaternized products.

Also suitable as copolymerizable monomers are acrylamidoglycolic acid, vinylsulfonic acid, allylsulfonic acid, methallylsulfonic acid, styrenesulfonic acid, 3-sulfopropyl acrylate, 3-sulfopropyl methacrylate and acrylamidomethylpropanesulfonic acid, and phosphono-containing monomers such as vinylphosphonic acid, allylphosphonic acid and acrylamidomethylpropanephosphonic acid. Further suitable copolymerizable monomers are N-vinylpyrrolidone, N-vinylimidazole, N-vinyl-2-methylimidazole, N-vinyl-4-methylimidazole, diallylammonium chloride and vinyl esters such as vinyl acetate and vinyl propionate, and vinylaromatic compounds such as styrene. It is, of course, also possible to employ mixtures of said monomers.

The homo- and copolymers are prepared by known processes, eg. of solution, precipitation, suspension or inverse suspension polymerization, or of emulsion or inverse emulsion polymerization, using compounds which form free radicals under the polymerization conditions.

45

The polymerizations are normally carried out at from 30 to 200°C, preferably 40 to 110°C. Examples of suitable initiators are azo

and peroxy compounds, and the conventional redox initiator systems such as combinations of hydrogen peroxide and reducing compounds, eg. sodium sulfite, sodium bisulfite, sodium formaldehyde sulfoxylate and hydrazine.

5

The copolymers have K values of at least 7, preferably 10 to 100, particularly preferably 10 to 50. The K values are determined via the method of H. Fikentscher, Cellulose-Chemie, 13 (1932) 58-64 and 71-74, in aqueous solution or in an organic solvent at 25°C, 10 at concentrations which are from 0.1% to 5%, depending on the K value range.

Besides the polymeric binders described above, it is possible to employ in particular up to 30% by weight, based on the total 15 weight of the binder, of other binders such as polymers, copolymers, cellulose derivatives, starch and starch derivatives. Suitable examples are:

Polyvinylpyrrolidone (PVP), copolymers of N-vinylpyrrolidone 20 (NVP) and vinyl acetate or vinyl propionate, copolymers of vinyl acetate and crotonic acid, partially hydrolyzed polyvinyl acetate, polyvinyl alcohol, poly(hydroxyalkyl acrylates), poly(hydroxyalkyl methacrylates), polyacrylates and polymethacrylates (Eudragit types), copolymers of methyl 25 methacrylate and acrylic acid, polyacrylamides, polyethylene glycols, polyvinylformamide (partially or completely hydrolyzed where appropriate), cellulose esters, cellulose ethers, especially methyl cellulose and ethyl cellulose, hydroxyalkylcelluloses, especially hydroxypropylcellulose, 30 hydroxyalkylalkylcelluloses, especially hydroxypropylethylcellulose, cellulose phthalates, especially cellulose acetate phthalate and hydroxypropylmethylcellulose phthalate, and mannans, especially galactomannans. Of these, polyvinylpyrrolidone, copolymers of N-vinylpyrrolidone and vinyl 35 esters, poly(hydroxyalkyl acrylates), poly(hydroxyalkyl methacrylates), polyacrylates, polymethacrylates, alkylcelluloses and hydroxyalkylcelluloses are particularly preferred.

The polymeric binder must soften or melt in the complete mixture 40 of all the components in the range of from 50 to 180°C, preferably 60 to 130°C, to form a plastic mixture. The glass transition temperature of the mixture must therefore be below 180°C, preferably below 130°C. If necessary, it is reduced by conventional pharmacologically acceptable plasticizing 45 auxiliaries. The amount of plasticizer does not exceed 30% of the total weight of binder and plasticizer in order to form storage-stable drug forms which show no cold flow. However, the

mixture preferably contains no plasticizer.

Examples of such plasticizers are:

- 5 Long-chain alcohols, ethylene glycol, propylene glycol, glycerol, trimethylolpropane, triethylene glycol, butanediols, pentanols such as pentaerythritol, hexanols, polyethylene glycols, polypropylene glycols, polyethylene/propylene glycols, silicones, aromatic carboxylic esters (eg. dialkyl phthalates, trimellitic
10 esters, benzoic esters, terephthalic esters) or aliphatic dicarboxylic esters (eg. dialkyl adipates, sebacic esters, azelaic esters, citric and tartaric esters), fatty acid esters such as glycerol mono-, di- or triacetate or sodium diethyl sulfosuccinate. The concentration of plasticizer is generally
15 from 0.5 to 15, preferably 0.5 to 5, % of the total weight of the mixture.

- Conventional pharmaceutical auxiliaries, whose total amount can be up to 100% of the weight of the polymer, are, for example,
20 extenders and bulking agents such as silicates or diatomaceous earth, magnesium oxide, aluminum oxide, titanium oxide, stearic acid or its salts, eg. the magnesium or calcium salt, methylcellulose, sodium carboxymethylcellulose, talc, sucrose, lactose, cereal or corn starch, potato flour, polyvinyl alcohol,
25 in particular in a concentration of from 0.02 to 50, preferably 0.20 to 20, % of the total weight of the mixture.

Lubricants such as aluminum and calcium stearates, talc and silicones, in a concentration of from 0.1 to 5, preferably 0.1 to
30 3, % of the total weight of the mixture.

Flowability agents such as animal or vegetable fats, especially in hydrogenated form and those which are solid at room temperature. These fats preferably have a melting point of 50°C or
35 above. Triglycerides of C₁₂, C₁₄, C₁₆ and C₁₈ fatty acids are preferred. It is also possible to use waxes such as carnauba wax. These fats and waxes may be admixed advantageously alone or together with mono- and/or diglycerides or phosphatides, especially lecithin. The mono- and diglycerides are preferably
40 derived from the abovementioned fatty acid types. The total amount of fats, waxes, mono-, diglycerides and/or lecithins is from 0.1 to 30, preferably 0.1 to 5, % of the total weight of the composition for each layer.

45 Dyes, such as azo dyes, organic or inorganic pigments or dyes of natural origin, with preference for inorganic pigments in a concentration of from 0.001 to 10, preferably 0.5 to 3, % of the

total weight of the mixture.

Stabilizers such as antioxidants, light stabilizers, hydroperoxide destroyers, radical scavengers, stabilizers against 5 microbial attack.

It is also possible to add wetting agents, preservatives, disintegrants, adsorbents, release agents and propellants (cf., for example, H. Sucker et al., Pharmazeutische Technologie, 10 Thieme-Verlag, Stuttgart 1978).

Auxiliaries include for the purpose of the invention substances for producing a solid solution of the active ingredient. Examples of these auxiliaries are pentaerythritol and pentaerythritol 15 tetraacetate, polymers such as polyethylene oxides and polypropylene oxides and their block copolymers (poloxamers), phosphatides such as lecithin, homo- and copolymers of vinylpyrrolidone, surfactants such as polyoxyethylene 40 stearate, and citric and succinic acids, bile acids, sterols 20 and others as indicated, for example, in J. L. Ford, Pharm. Acta Helv. 61 (1986) 69-88.

Auxiliaries are also regarded as being bases and acids added to control the solubility of an active ingredient (see, for example, 25 K. Thoma et al., Pharm. Ind. 51 (1989) 98-101).

The only precondition for the suitability of auxiliaries is adequate thermal stability.

30 Active ingredients mean for the purpose of the invention all substances with a physiological effect as long as they do not decompose under the processing conditions. These are, in particular, pharmaceutical active ingredients (for humans and animals), active ingredients for plant treatment, insecticides, 35 active ingredients of human and animal foods, fragrances and perfume oils. The amount of active ingredient per dose unit and the concentration may vary within wide limits depending on the activity and the release rate. The only condition is that they suffice to achieve the desired effect. Thus, the concentration of 40 active ingredient can be in the range from 0.1 to 95, preferably from 20 to 80, in particular 30 to 70, % by weight. It is also possible to employ combinations of active ingredients. Active ingredients for the purpose of the invention also include vitamins and minerals. The vitamins include the vitamins of the A 45 group, the B group, by which are meant besides B₁, B₂, B₆ and B₁₂ and nicotinic acid and nicotinamide also compounds with vitamin B properties such as adenine, choline, pantothenic acid, biotin,

adenylic acid, folic acid, orotic acid, pangamic acid, carnitine, p-aminobenzoic acid, myo-inositol and lipoic acid, and vitamin C, vitamins of the D group, E group, F group, H group, I and J groups, K group and P group. Active ingredients for the purpose
5 of the invention also include therapeutic peptides. Plant treatment agents include, for example, vinclozolin, epoxiconazole and quinmerac.

The novel process is suitable, for example, for processing the
10 following active ingredients:

acebutolol, acetylcysteine, acetylsalicylic acid, aciclovir,
alprazolam, alfacalcidol, allantoin, allopurinol, ambroxol,
amikacin, amiloride, aminoacetic acid, amiadarone, amitriptyline,
15 amlodipine, amoxicillin, ampicillin, ascorbic acid, aspartame,
astemizole, atenolol, beclomethasone, benserazide, benzalkonium hydrochloride, benzocaine, benzoic acid, betamethasone,
bezafibrate, biotin, biperiden, bisoprolol, bromazepam,
bromhexine, bromocriptine, budesonide, bufexamac, buflomedil,
20 buspirone, caffeine, camphor, captopril, carbamazepine,
carbidopa, carboplatin, cefachlor, cefalexin, cefadroxil,
cefazoline, cefixime, cefotaxime, ceftazidime, ceftriaxone,
cefuroxime, selegiline, chloramphenicol, chlorhexidine,
chlorpheniramine, chlortalidone, choline, cyclosporin,
25 cilastatin, cimetidine, ciprofloxacin, cisapride, cisplatin,
clarithromycin, clavulanic acid, clomipramine, clonazepam,
clonidine, clotrimazole, codeine, cholestyramine, cromoglycic acid, cyanocobalamin, cyproterone, desogestrel, dexamethasone,
dexpanthenol, dextromethorphan, dextropropoxiphenone, diazepam,
30 diclofenac, digoxin, dihydrocodeine, dihydroergotamine,
dihydroergotoxin, diltiazem, diphenhydramine, dipyridamole,
dipyrrone, disopyramide, domperidone, dopamine, doxycycline,
enalapril, ephedrine, epinephrine, ergocalciferol, ergotamine,
erythromycin, estradiol, ethinylestradiol, etoposide, Eucalyptus
35 globulus, famotidine, felodipine, fenofibrate, fenoterol,
fentanyl, flavin mononucleotide, fluconazole, flunarizine,
fluorouracil, fluoxetine, flurbiprofen, furosemide, gallopamil,
gemfibrozil, gentamicin, Gingko biloba, glibenclamide, glipizide,
clozapine, Glycyrrhiza glabra, griseofulvin, guaifenesin,
40 haloperidol, heparin, hyaluronic acid, hydrochlorothiazide,
hydrocodone, hydrocortisone, hydromorphone, ipratropium hydroxide, ibuprofen, imipenem, indomethacin, iohexol, iopamidol,
isosorbide dinitrate, isosorbide mononitrate, isotretinoin,
ketotifen, ketoconazole, ketoprofen, ketorolac, labetalol,
45 lactulose, lecithin, levocarnitine, levodopa, levoglutamide,
levonorgestrel, levothyroxine, lidocaine, lipase, imipramine,
lisinopril, loperamide, lorazepam, lovastatin,

- medroxyprogesterone, menthol, methotrexate, methyldopa,
methylprednisolone, metoclopramide, metoprolol, miconazole,
midazolam, minocycline, minoxidil, misoprostol, morphine,
multivitamin mixtures or combinations and mineral salts,
- 5 N-methylephedrine, naftidrofuryl, naproxen, neomycin,
nicardipine, nicergoline, nicotinamide, nicotine, nicotinic acid,
nifedipine, nimodipine, nitrazepam, nitrendipine, nizatidine,
norethisterone, norfloxacin, norgestrel, nortriptyline, nystatin,
ofloxacin, omeprazole, ondansetron, pancreatin, panthenol,
- 10 pantothenic acid, paracetamol, penicillin G, penicillin V,
phenobarbital, pentoxifylline, phenoxymethylenicillin,
phenylephrine, phenylpropanolamine, phenytoin, piroxicam,
polymyxin B, povidone-iodine, pravastatin, prazepam, prazosin,
prednisolone, prednisone, bromocriptine, propafenone,
- 15 propranolol, proxyphylline, pseudoephedrine, pyridoxine,
quinidine, ramipril, ranitidine, reserpine, retinol, riboflavin,
rifampicin, rutoside, saccharin, salbutamol, salcatorin,
salicylic acid, simvastatin, somatropin, sotalol, spironolactone,
sucralfate, sulbactam, sulfamethoxazole, sulfasalazine,
- 20 sulpiride, tamoxifen, tegafur, teprenone, terazosin, terbutaline,
terfenadine, tetracycline, theophylline, thiamine, ticlopidine,
timolol, tranexamic acid, tretinoin, triamcinolone acetonide,
triamterene, trimethoprim, troxerutin, uracil, valproic acid,
vancomycin, verapamil, vitamin E, folinic acid, zidovudine.
- 25 Preferred active ingredients are ibuprofen (as racemate,
enantiomer or enriched enantiomer), ketoprofen, flurbiprofen,
acetylsalicylic acid, verapamil, paracetamol, nifedipine or
captopril.
- 30 To produce the solid dosage forms, a plastic mixture of the
components (melt) is prepared and then subjected to a shaping
step. There are various ways of mixing the components and forming
the melt. The mixing can take place before, during and/or after
35 the formation of the melt. For example, the components can be
mixed first and then melted or be mixed and melted
simultaneously. The plastic mixture is often then homogenized in
order to disperse the active ingredient thoroughly.
- 40 However, it has proven preferable, especially when sensitive
active ingredients are used, first to melt the polymeric binder
and, where appropriate, make a premix with conventional
pharmaceutical additives, and then to mix in (homogenize) the
sensitive active ingredient(s) in the plastic phase in intensive
45 mixers with very short residence times. The active ingredient(s)
can for this purpose be employed in solid form or in solution or
dispersion.

10

The components are generally employed as such in the production process. However, they can also be used in liquid form, ie. as solution, suspension or dispersion.

- 5 Suitable solvents for the liquid form of the components are primarily water or a water-miscible organic solvent or a mixture thereof with water. However, it is also possible to use organic solvents which are immiscible or miscible with water. Suitable water-miscible solvents are, in particular, C₁-C₄-alkanols such as
- 10 ethanol, isopropanol or n-propanol, polyols such as ethylene glycol, glycerol and polyethylene glycols. Suitable water-immiscible solvents are alkanes such as pentane or hexane, esters such as ethyl acetate or butyl acetate, chlorinated hydrocarbons such as methylene chloride, and aromatic
- 15 hydrocarbons such as toluene and xylene. Another solvent which can be used is liquid CO₂.

The solvent used in the individual case depends on the component to be taken up and the properties thereof. For example,

- 20 pharmaceutical active ingredients are frequently used in the form of a salt which is, in general, soluble in water. Water-soluble active ingredients can therefore be employed as aqueous solution or, preferably, be taken up in the aqueous solution or dispersion of the binder. A corresponding statement applies to active
- 25 ingredients which are soluble in one of the solvents mentioned, if the liquid form of the components used is based on an organic solvent.

It is possible where appropriate to replace melting by

- 30 dissolving, suspending, or dispersing in the abovementioned solvents, if desired and/or necessary with the addition of suitable auxiliaries such as emulsifiers. The solvent is then generally removed to form the melt in a suitable apparatus, eg. an extruder. This will be comprised by the term mixing
- 35 hereinafter.

The melting and/or mixing takes place in an apparatus customary for this purpose. Particularly suitable ones are extruders or containers which can be heated where appropriate and have an

40 agitator, eg. kneaders (like those of the type to be mentioned below).

A particularly suitable mixing apparatus is one employed for mixing in plastics technology. Suitable apparatuses are

45 described, for example, in "Mischen beim Herstellen und Verarbeiten von Kunststoffen", H. Pahl, VDI-Verlag, 1986. Particularly suitable mixing apparatuses are extruders and

11

dynamic and static mixers, and stirred vessels, single-shaft stirrers with stripper mechanisms, especially paste mixers, multishaft stirrers, especially PDSM mixers, solids mixers and, preferably, mixer/kneader reactors (eg. ORP, CRP, AP, DTB

5 supplied by List or Reactotherm supplied by Krauss-Maffei or Ko-Kneter supplied by Buss), trough mixers and internal mixers or rotor/stator systems (eg. Dispax supplied by IKA).

In the case of sensitive active ingredients it is preferable
10 first for the polymeric binder to be melted in an extruder and then for the active ingredient to be admixed in a mixer/kneader reactor. On the other hand, with less sensitive active ingredients, a rotor/stator system can be employed for vigorously dispersing the active ingredient.

15

The mixing apparatus is charged continuously or batchwise, depending on its design, in a conventional way. Powdered components can be introduced in a free feed, eg. via a weigh feeder. Plastic compositions can be fed in directly from an
20 extruder or via a gear pump, which is particularly advantageous if the viscosities and pressures are high. Liquid media can be metered in by a suitable pump unit.

The mixture obtained by mixing and/or melting the binder, the
25 active ingredient and, where appropriate, the additive(s) ranges from pasty to viscous (plastic) or fluid and is therefore extrudable. The glass transition temperature of the mixture is below the decomposition temperature of all the components present in the mixture. The binder should preferably be soluble or
30 swellable in a physiological medium.

The steps of mixing and melting in the process can be carried out in the same apparatus or in two or more separately operating apparatuses. The preparation of a premix can take place in one of
35 the conventional mixing apparatuses described above. A premix of this type can then be fed directly, for example, into an extruder and subsequently extruded, where appropriate with the addition of other components.

40 It is possible in the novel process to employ as extruders single screw machines, intermeshing screw machines or else multiscrew extruders, especially twin screw extruders, corotating or counterrotating and, where appropriate, equipped with kneading disks. If it is necessary in the extrusion to evaporate a
45 solvent, the extruders are generally equipped with an evaporating section. Particularly preferred extruders are those of the ZKS series from Werner & Pfleiderer.

It is also possible according to the invention to produce multilayer pharmaceutical forms by coextrusion, in which case a plurality of mixtures of the components described above is fed 5 together to an extrusion die so as to result in the required layered structure of the multilayer pharmaceutical form. It is preferable to use different binders for different layers.

Multilayer drug forms preferably comprise two or three layers. 10 They may be in open or closed form, in particular as open or closed multilayer tablets.

At least one of the layers contains at least one pharmaceutical active ingredient. It is also possible for another active 15 ingredient to be present in another layer. This has the advantage that two mutually incompatible active ingredients can be processed or that the release characteristics of the active ingredient can be controlled.

20 The shaping takes place by coextrusion with the mixtures from the individual extruders or other units being fed into a common coextrusion die and extruded. The shape of the coextrusion die depends on the required pharmaceutical form. Examples of suitable dies are those with a flat orifice, called a slit die, and dies 25 with an annular orifice. The design of the die depends on the polymeric binder used and the required pharmaceutical form.

The resulting mixture is preferably solvent-free, ie. it contains neither water nor an organic solvent.

30 The plastic mixture is, as a rule, subjected to final shaping. This can result in a large number of shapes depending on the die and mode of shaping. For example, if an extruder is used, the extrudate can be shaped between a belt and a roll, between two 35 belts or between two rolls, as described in EP-A-358 105, or by calendering in a calender with two molding rolls, see, for example, EP-A-240 904. Other shapes can be obtained by extrusion and hot- or cold-cut of the extrudate, for example small-particle and uniformly shaped pellets. Hot-cut pelletization usually 40 results in lenticular dosage forms (tablets) with a diameter of from 1 to 10 mm, while strip pelletization normally results in cylindrical products with a length to diameter ratio of from 1 to 10 and a diameter of from 0.5 to 10 mm. It is thus possible to produce monolayer but also, on use of coextrusion, open or closed 45 multilayer dosage forms, for example oblong tablets, coated tablets, pastilles and pellets. The resulting granules can also be ground to a powder and compressed to tablets in a conventional

13

way. Micropastilles can be produced by the Rotoform-Sandvik process. These dosage forms can be rounded and/or provided with a coating by conventional methods in a subsequent process step.

Examples of materials suitable for film coatings are

5 polyacrylates such as the Eudragit types, cellulose esters such as the hydroxypropylcellulose phthalates, and cellulose ethers, such as ethylcellulose, hydroxypropylmethylcellulose or hydroxypropylcellulose.

10 In specific cases there may be formation of solid solutions. The term solid solutions is familiar to the skilled worker, for example from the literature cited at the outset. In solid solutions of active ingredients in polymers, the active ingredient is in the form of a molecular dispersion in the
15 polymer.

The following examples are intended to illustrate the novel process without restricting it, however.

20 Examples

Example 1

520 g of polyvinylcaprolactam (K value 40; 1% strength in
25 ethanol) were extruded with 480 g of verapamil hydrochloride under the conditions indicated below and calendered to give 500 mg oblong tablets.

30	Section 1	52°C
	Section 2	85°C
	Section 3	130°C
	Section 4	108°C
	Section 5	92°C
	Die	85°C

35

The release after 1 hour was 100% (USP paddle method (pH change)).

Example 2

40 500 g of copolymer of 50% by weight vinylcaprolactam and 50% by weight vinylpyrrolidone (K value 65; 1% strength in water) were extruded with 480 g of verapamil hydrochloride under the conditions indicated below to give 500 mg oblong tablets, and
45 were granulated.

Section 1 47°C

14

	Section 2	86°C
	Section 3	132°C
	Section 4	112°C
	Section 5	101°C
5	Die	100°C

The release after 1 hour was 67% and after 2 hours was 100% (USP paddle method (pH change)).

10 Example 3

500 g of copolymer of 50% by weight vinylcaprolactam and 50% by weight vinylpyrrolidone (K value 65; 1% strength in water) were extruded with 500 g of vinclozolin under the conditions indicated **15** below, and were cooled and granulated.

	Section 1	60°C
	Section 2	96°C
	Section 3	140°C
20	Section 4	150°C
	Section 5	129°C
	Die	100°C

Transparent, X-ray-amorphous, water-dispersible granules were **25** obtained.

Example 4

500 g of polyvinylcaprolactam (K value 40; 1% strength in **30** ethanol) were extruded with 500 g of epoxyconazole under the conditions indicated below, and were cooled and calendered.

	Section 1	60°C
	Section 2	90°C
35	Section 3	120°C
	Section 4	135°C
	Section 5	120°C
	Die	100°C

40 Transparent, X-ray-amorphous, water-dispersible granules were obtained.

45

We claim:

1. A process for producing solid dosage forms by mixing at least
5 one polymeric binder, at least one active ingredient and,
where appropriate, conventional additives to form a plastic
mixture, and shaping, wherein a homo- and/or copolymer of
N-vinylcaprolactam is used as polymeric binder.
- 10 2. A process as claimed in claim 1, wherein a homopolymer of
N-vinylcaprolactam is used as polymeric binder.
- 15 3. A process as claimed in claim 1, wherein a copolymer of
N-vinylcaprolactam which comprises at least 10% by weight, in
particular at least 25% by weight and, particularly
preferably, at least 50% by weight of N-vinylcaprolactam
units is used as polymeric binder.
- 20 4. A process as claimed in claim 1 or 3, wherein a copolymer of
N-vinylcaprolactam which comprises as comonomer at least one
copolymizable monomer which is selected from monoethylenically
unsaturated mono- and dicarboxylic acids having 3 to 30
carbon atoms; diesters and monoesters of these carboxylic
acids with C₁-C₂₄-alkanols, C₂-C₄-alkanediois and mono- or
25 di-C₁-C₄-alkylamino-C₂-C₄-alkanols; amides, mono- and
di-C₁-C₄-alkylamides and nitriles of these carboxylic acids,
vinyl esters of aliphatic C₁-C₁₈-carboxylic acids, and
N-vinylpyrrolidone and the salts thereof, is used as
polymeric binder.
- 30 5. A process as claimed in claim 4, wherein the comonomer is
selected from monoethylenically unsaturated C₃-C₆-mono- and
dicarboxylic acids and the esters of these carboxylic acids
with C₁-C₁₈-alkanols and C₂-C₄-alkanediois.
- 35 6. A process as claimed in claim 5, wherein the
comonomer is selected from acrylic acid, methacrylic acid,
maleic acid, esters of acrylic acid or methacrylic acid with
methanol, ethanol, n-butanol, 2-ethylhexanol, ethylene glycol
40 and propylene glycol and, in particular, N-vinyl-
pyrrolidone.
- 45 7. A process as claimed in any of the preceding claims, wherein
the polymeric binder has a K value in the range from 10 to
100.
8. A process as claimed in any of the preceding claims, wherein

2

the plastic mixture is produced by mixing and/or melting the components in an extruder.

9. A process as claimed in any of the preceding claims for
5 producing pharmaceutical dosage forms, plant treatment compositions, animal feed additives and supplements, human food supplements and dosage forms for fragrances and perfume oils.

10 10. A solid dosage form obtainable by a process as claimed in any of claims 1 to 9.

15

20

25

30

35

40

45